



STATUS OF GROUNDWATER FLUORIDE IN SOME AREAS OF SANGANER TEHSIL OF JAIPUR DISTRICT, RAJASTHAN (INDIA)

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Abstract

Fluoride in groundwater causes genuine medical conditions and heaps of illnesses in numerous pieces of India. Jaipur has as of late been distinguished together of the savvy urban areas and one among numerous tehsils. Those the high fluoride content was incited to reconsider the subject and rethink the fluoride content status. Considering this, we have attempted to work out the amount of fluoride and pollution inside the groundwater parts of Sanganer Tehsil District - Jaipur, Rajasthan. Water quality in parched and semi-dry zones has been declining in the course of recent many years. Sanganer Tehsil is fighting the matter of groundwater contamination. Inside the momentum research, fluoride (F) passable fluoride content ($> 1.5 \text{ mg L}^{-1}$) in refreshment chose in 23 towns in Sanganer tehsil (115 examples were gathered from the space). Water tests are basic with a pH somewhere in the range of 7.56 and 9.78. Electric conductivity (EC) goes from 312 mhos cm^{-1} to 4562 mhos cm^{-1} . Calcium hardness (ca-h) is among 10 and a hundred ninety mg l^{-1} . General hardness (T.H.) shifts among forty and 475 mg l^{-1} . Chloride goes from forty-five.00 mg l^{-1} to 710 mg l^{-1} and fluoride from zero.30 to 6.42 mg l^{-1} . The outcomes showed that the groundwater top-notch of a couple of districts of Sanganer tehsil was truly horrendous, undeserving for drinking, and will best be utilized after the correct cure.

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Introduction

Ecological contamination by fluoride can have serious ramifications for life forms, vegetation, and fauna because of water, soil, air, or a combination of each of the three mankind. The matter of high fluoride inside the climate in India and in this manner the subsequent wellbeing risks have been uncovered. The size of the worth is important to investigate factors like the wellspring of fluoride, the topographical appropriation of fluoride, and its scattering [Rajesh Kumar Yadav et al. 2012]. Soil and water assets are being dirtied on account of different natural elements like unsafe squanders, fluid and soil squanders from businesses, removal, surface hindrances, and so on By and large, water assets are undependable for human utilization, yet in addition for different exercises related with water system and mechanical requirements. The presence of fluoride in high fixations in food and water turned into an overall concern. Around 200 million individuals from 25 nations experience the ill effects of the staggering impacts of fluorosis Malti Rotwar et al. 2020]. Fluoride might be a useful normal supplement found in differing focuses in air, water, and soil. At the point when devoured in ideal sums, it improves dental wellbeing, however overabundance consumption (>1.5 mg-1) may cause fluorosis, including dental, skeletal, and non-skeletal fluorosis optional neurological confusions [Indermitte, E. et al. 2007 and Shailaja, K. and M.E.C. Jhonson, 2007]. Reliable with safe drinking quality guidelines, the fluoride focus ought to be inside the scope of 1.00 to 1.50 mg-1 and past the upper level; it brings about destructive impacts on the body (WHO, 1984).

Fluorine is a most plentiful component inside the and ought to enter groundwater inside the fluoride (F-), a geochemical with fluoride-containing minerals like mycoses, hornblende,

pyroxenes a consequence of the discussion by air precipitation (Jacks, G et al. 2005). High groupings of fluoride in groundwater are typically shakes in soluble groundwater - treated with low calcium or bicarbonate pop (Edmunds, W. M.. and Smedley, P. L., 2013). It additionally influences fluoride dissolvability (Edmunds, W. M., Smedley, P. L., 2013 and Ayoob, S.; Gupta, A. K 2006). For the most part, carbonate rocks go about as zinc of fluoride, and the deliberate fluoride focuses are equivalent to soil pH (Bishnoi, M et al. 2007, Chandrasekhram, D. et al. 2008, Gupta, S.C.,1991, Handa, B.K., 1988). Increased fluoride levels in groundwater because of an expanded alkalinity in soils. Anthropogenic materials likewise assume a part in the utilization of phosphate-based composts, which regularly contain significant degrees of fluoride. (Chaney, R. L., 2012).

The danger of fluorosis is expanding quickly around the world. India faces comparable difficulties. Presently, fluoride is discovered to be high in 17 Indian locales, including Rajasthan. In Rajasthan, 22 provinces (32) presently utilize more fluoride than is permitted. Past activists (Bishnoi, M. and S. Arora 2018, Yadav, A.K., et al., 2003) have detailed that fluoride and fluorosis in refreshments are related with a high centralization of fluoride particles. Typically, surface water doesn't contain high fluoride; however groundwater is sullied with high fluoride content in light of the fact that the primary wellsprings of fluoride will be fluoride-containing rocks. As water goes through rocks, fluoride escapes from these stones. Subsequently, it is a wellspring of fluoride in India.

Groundwater is the start of the surface. Rocks contain up to 180 g-1 of sand in sandstone and up to 800 g/1 of stone (Sharma, D.K., 1990). In our

investigation region, generally 40% of the populace needs refreshment from groundwater sources like hand siphons and open wells. Ingestion of antagonistic impacts of fluoride on the body liquids, particularly in the teeth and bones, just as the far and wide event of fluorosis of teeth and bones. A study was directed inside the examination region. Groundwater tests were gathered from 23 towns and towns from the investigation region.

Study Area

Sanganer Tehsil is an examination region situated inside the Jaipur District of Rajasthan, India. The geographic space of Jaipur locale is 11,061.44 sq km and stretches between 26° 25' north scopes. Moreover, 27° 51' and east longitude 74° 55' and 76° 15' cover the east-focal piece of Rajasthan, isolating 13 areas into tehsils and 13 squares for managerial accommodation. Sanganer Tehsil is associated with the downtown area of Jaipur. Its elevation goes from 26° 49'N to 26° 51'N. It is between 75° 46'E and 75° 51'E. Its territory is 635.5 ch. Km, the environment of this district is semi-dry and warm with power. Temperature (15–45 °C) and precipitation 650 mm (26 inches). The number of inhabitants in Sanganer Tehsil is 573171 according to the 2011 registration.

Methodology

Groundwater tests (open wells/hand siphons) are gathered at the 23 towns and states from Sanganer tehsil destinations as given in Table 1. 115 water tests (5 examples for every district) were gathered in pre-purged polyethylene. 1-liter container. The examining occurred arbitrarily in July and August 2015. Water tests were shipped off the research center for investigation utilizing standard Physico-synthetic scale strategies. The estimation is resolved as fluoride (F) with the assistance of a particular particle meter (Fluoride Analysis-Fluoride focuses in water tests)

dissected with the Orion Research Ion Analyzer Model 407 A, a fluoride particle specific cathode.). Common systems are followed to recognize fluoride focuses (APHA 2005). The ionic concentrate (TISAB) used to keep up adequate ionic energy and to stay away from the mind-boggling development of good outcomes. Likewise, Physico-compound boundaries, for example, pH, EC, C-hardness, complete hardness, chloride, and alkalinity are additionally assessed to be predictable with regular techniques (APHA 2005) the comparing quality refreshment costs (APHA 2005, USPH 1985, WHO 1996, I.S.I. 1982 are given in Table 2.

Results and Discussion

Examination results for the different examples gathered in the investigation territory (23 towns) in Sanganer tehsil are introduced in Table 1. The work uncovered that fluoride tests of groundwater tests in fourteen areas changed from 1.5-6.42 mg l-1. In three districts, fluoride focus went from 5.0-6.42 mg l-1. The convergence of fluoride in three spots is exceptionally disturbing. The most extreme convergence of fluoride was recorded in Pratap Nagar Area 6.42 mg l-1. The passable furthest reaches of fluoride focus is 1-1.5 ppm, comparing to (WHO 1996). data have shown that hourly zones in the tehsil study territory are influenced by high fluoride focus, with 40% underneath as far as possible. (Fig. 4). The fluoride fixation distinction is most likely because of the distinction in the synthetic line of the stone. pH is communicated as the reach from 0-10. esteem is an impression of the convergence of H+ particles inside the arrangement. The pH esteem comparable to the examination territory was found inside grades 7.56 to 9.78. The complete pH esteem was found in the Mahal test (9.78), and the low pH (7.56) was seen from Maheshnagar (Table 1 and Fig. 1)., In understanding with WHO (1996). pH was

between 6.9-9.2. it has been seen that pH created an immediate collaboration with fluoride filtration, demonstrating that high water levels advance fluoride spillage and hence influence groundwater (Teotia, S.P.S., et al. 1981, Wodeyar, B.K and G. Sreenivasan 1996). Fluoride fixation is additionally identified with alkalinity (Trivedi, P., 1988). Power productivity can be a value impression of the capacity of the current force catch arrangement. The U.S. suggested the admissible power age (EC) cutoff of 300 $\mu\text{mhoS cm}^{-1}$ (Table 2). Least (312 $\mu\text{mhoS cm}^{-1}$) and greatest (2365 $\mu\text{mhoS cm}^{-1}$) E.C. were accounted for from Mangyawas and Manipura, individually (Table 1 and Fig.3). An immediate connection was seen between E.C. also, F as prior announced (Devi, S., et al. 2003). Water hardness isn't a choice yet maybe an adaptable and complex combination of cations and anions. The principal particle causing particles are calcium and magnesium. Calcium focus (Ca-H) goes from 10 to 190 mg l^{-1} . A modest quantity of Ca-H (10 mg l^{-1}) was seen from Chimanpura, while a lot of Ca-H (190 mg l^{-1}) was accounted for from Gopalpura (Table 1 Fig. 6). Complete hardness (TH) fluctuates from 40 to 475 mg l^{-1} . Least (40 mg l^{-1}) and measurement (475 mg l^{-1}) were accounted for from Gopalpura and Chimanpura, separately (Table 1 and Fig. 5). The WHO suggested a protected constraint of hardness, i.e., 100-500 mg l^{-1} (Table 2). In spring water, the hardness is fundamental because of carbonates, carbonates, sulfates, and chlorides of Ca and Mg. The outcomes are predictable with the discoveries of (28). This generally because of low liquefying fluoride (29). Chloride changed from 45 to 710 mg l^{-1} . The least (45 mg l^{-1}) was accounted for from Ramnagarya, and the greatest (710 mg l^{-1}) was seen from Gopalpura (Table 1 and Fig. 7). The higher substance of chlorides gives a pungent taste to water. Alkalinity went from Minimum (289 mg l^{-1}) Durgapura to most

extreme (2250 mg l^{-1}) from Jaipur (Table 1 and Fig. 8). The high worth of alkalinity gives an undesirable taste to water. It showed an immediate relationship with pH, F, E.C. The outcomes are in concurrence with the aftereffects of (Jain, P. et al. 2005, Hem, J.D. 1991).

The information shows that the example of groundwater gathered at different Sanganer Tehsil locales is seriously debated by undeniable degrees of fluoride and alkalinity, which is a significant danger to human wellbeing. Numerous boundaries were completely or mostly permitted. Along these lines, it was inferred that fluoride penetration principally by groundwater affected tooth and bone arrangement. In this manner, the neighborhood drink of Sanganer Tehsil can't be cooked. To deal with spring water quality, persistent preparation of Physico-compound boundaries ought to be performed. It must be utilized for cooking and drinking after past treatment. The creators unequivocally suggest that quick advances be taken for water removal, for example, The Nalgonda cycle, created by the National Environmental Engineering Research Institute. It is an interaction that includes a fast blending of water with lime (sodium or calcium carbonate), alum (aluminum sulfate), and chlorinated lime. This prompts slants and landslides. Thusly, the supernatant, which may contain just adequate measures of fluoride, is typically sifted or weakened and utilized for cooking and drinking. A few majors of nutrient C limitation on high-fat weight control plans, drink a lot of milk and eat calcium-rich vegetables like verdant vegetables. In the event that any indications of fluorosis are distinguished, stay away from the fundamental wellsprings of fluoride consumption.

References:

1. Rajesh Kumar Yadav, S. Sharma, M. Bansal, A. Singh, V. Panday, R. Maheshwari (2012). Effects of Fluoride Accumulation on Growth of Vegetables and Crops in Dausa District, Rajasthan, India. *Advances in Bioresearch*. 3 (4) :14-16.
2. Malti Rotwar, Gaurav Sharma, Rajesh Kumar Yadav, Yashoda Saini, (2020). Accretion of Fluoride Gist in Foodstuff of Fluoride Overstresses Area of Sanganer Tehsil, Jaipur District - Rajasthan (India). *European Journal of Molecular & Clinical Medicine*. 7(11): 2001-2006
3. Indermitte, E., E. Karro and A. Saava. (2007): Tap water fluoride levels in Estonia. *Fluoride*, 40, 244-247
4. Shailaja, K. and M.E.C. Jhonson (2007): Fluorides in groundwater and its impact on health. *J. Environ. Biol.*, 28, 331-332
5. WHO (1984): Fluorine and fluoride? *Environmental Health Criteria*. Geneva, 36, 93
6. Jacks, G.; Bhattacharya, P.; Chaudhary, V.; Singh, K. Controls on the genesis of some high-fluoride groundwaters in India. *Appl. Geochem.* 2005, 20 (2), 221-228.
7. Edmunds, W. M.; Smedley, P. L. Fluoride in natural waters. In *Essentials of Medical Geology*; Springer: Dordrecht, 2013; pp 311-336.
8. Ayoob, S.; Gupta, A. K. Fluoride in drinking water: a review on the status and stress effects. *Crit. Rev. Environ. Sci. Technol.* 2006, 36 (6), 433-487.
9. Chaney, R. L. Food safety issues for mineral and organic fertilizers. In *Advances in Agronomy*; Academic Press, 2012; Vol. 117, pp 51-116
10. Bishnoi, M. and S. Arora: Potable groundwater quality in some villages of Haryana, India: Focus on fluoride. *J. Environ. Biol.*, 28, 291-294 (2007).
11. Chandrasekhran, D. and S. Saji: High fluoride groundwater of Karbi-Anglong district, Assam. *Int. Conference on Groundwater Dynamics and Global Change*. pp. 19-22 (2008).
12. Chinoy, N.J., E. Sequeria, M.V. Narayana, M. Mathews, W. Barot, P.R. Kandoi, and D.D. Jhala: A survey of fluoride in 90 endemic villages of Mehsana and Banaskantha districts of north Gujrat, India. *Fluoride*, 38, 224 (2005).
13. Gangal, R.K. (2005): Fluoride hazard of groundwater in the Jaipur district, Rajasthan, India, and methods to mitigate it. *Fluoride*, 38, 241.
14. Gupta, S.C. (1991): Chemical character of groundwater in Nagaur district, Rajasthan. *Ind. J. Environ. Hlth.*, 33, 341-349.
15. Handa, B.K. (1988): Fluoride occurrences in natural water in India and its significance. *BHU-Jal News*, 3, 21-24.
16. Sharma, J.D., P. Jain and S. Deepika (2005): Geological study of fluoride in groundwater of Sanganer Tehsil of Jaipur

- district, Rajasthan, India. Fluoride, 38, 249
17. Stanley, V.A., N. Ramesh, K.S. Pillai and P.B.K. Murthy (1997): Epidemiological survey of fluorosis in an encore, Madras. Proc. Acad. Environ. Biol., 6, 121-125
 18. Yadav, A.K., P.K. Jain and S. Lal (2003): Geochemical study of fluoride in groundwater of Behror tehsil of Alwar district (Rajasthan). Res. J. Chem. Environ., 7, 43-47
 19. Sharma, D.K., C.P.S. Chandel and C.M. Gupta (1990): Fluoride levels in all types of water from various sources in nearby villages around Jaipur. Int. Indian Water Works Association, 10, 121
 20. APHA (2005): Standard methods for the examination of water and wastewater. 21st Edn., American Public Health Association, New York, U.S.A. (2005).
 21. USPH (1985): Fluoridation Census. Department of Health and Human Services. Bethesda MD. pp. 28-32
 22. WHO (1996): Guidelines for drinking water quality? 2, 231.
 23. I.S.I. (1982): Draft Indian Standard Specification for Drinking Water DOC: C.D.C. 26,8626, New Delhi.
 24. Teotia, S.P.S., M. Teotia and M.K. Singh (1981): Hydrogeochemical aspects of endemic skeletal fluorosis in India. An epidemiological study. Fluoride, 14, 69-74.
 25. Wodeyar, B.K and G. Sreenivasan (1996): Occurrence of fluoride in the groundwater and its impact in Peddvankahalla basin, Bellary district, Karnataka- A preliminary study. Curr. Sci., 70, 71-74.
 26. Trivedi, P.(1988): Relationship between fluoride, total alkalinity, total hardness in groundwater of Pali district in an arid and semi-arid region of western Rajasthan. Proc. Natl. Acad. Sci., India, 58, 7-11.
 27. Devi, S., S.B. Barbuddhe, D. Hazel and C. Dolly (2003): Physico-chemical characteristics of drinking water at velsao, Goa. J. Environ. Biol., 13, 203-209.
 28. Jain, P., J.D. Sharma and P. Sharma (2005): Chemical analysis of drinking water of villages of Sanganer Tehsil, Jaipur District. Int. J. Environ. Sci. Tech., 2, 373-379
 29. Hem, J.D. (1991): Study and interpret the chemical characteristic of natural water, United States geological survey water supply paper 2254, 3rd Edn. Scientific Publishers, Jodhpur. p. 120

S.No	Name of village	pH	T.D.S.	E.C.	F	TH	Ca ⁺ H	cl-	TA
			ppm	(µmhos/Cm	(mg/l)	(mg/l)	(mg/l)	(mg/l)	(mg/l)
1	Sanganer	8.6±0.03	1312±12.67	2310± ±9.66	3.5±0.07	360±11.43	80±1.88	190±5.11	510±4.98
	Mangyawas	7.85±0.04	197±14.57	312±7.34	1.5±0.02	252±16.98	167±0.99	360±1.90	810±12.20
2	Gopalpura	7.97±0.09	345±11.65	567±11.34	2.09±0.06	475±14.66	190±1.76	710±2.87	1245±10.38
3	Jaitpura	8.67±0.11	613±17.23	876± 10.89	3.7± 0.05	260±17.33	180±0.88	80±1.98	2250±9.23
4	Shoupura	8.86±0.01	1265±11.87	2100± 13.12	0.98±0.11	210±10.43	80±0.78	210±.78	1050±13.88
	Surajpura	8.65±0.02	754±17.33	1450± 19.78	3.9± 0.17	178±18.09	80±1.29	115±2.87	610±4.24
5	Bambal	8.1± 0.12	812±12.46	1513± 14.32	0.67±0.12	220±15.56	60±0.23	110±4.01	390±6.98
6	Dehla bas	8.87±0.06	1964±16.87	3987± 17.56	0.81±0.14	200±18.32	80±0.89	150±2.78	1120±8.45
	Nagalya Bhatt	8.87±0.14	2410±10.29	4562± 13.98	5.1± 0.06	115±10.56	35±1.20	90±4.73	550±11.09
8	Chak Getore	8.9± 0.08	613±9.45	710±16.43	0.5±0.08	240±9.06	150±1.55	185±5.10	410±14.05
9	Durgapura	8.67±0.10	716±14.84	1178± 15.09	0.73±0.12	330±16.21	50±1.26	328±0.77	289±7.22
10	Khokabas	8.85± 0.9	680±10.35	2130±11.98	0.3± 0.13	260±5.78	80±1.23	50±0.93	467±3.87
11	Jagatpura	8.56± 0.5	597±16.19	786±10.78	0.55±0.07	240±12.34	125±1.90	135±2.04	356±11.96
12	Malviya Nagar	7.9± 0.02	432± 8.16	810± 12.98	0.58±0.12	261±17.99	113±1.17	290±3.91	380±10.65
14	Mahal	9.78±0.01	1310±11.84	360±8.93	2.89±0.18	95± 11.68	45±0.78	85±1.88	1260±10.33
15	Ram nagarya	9.67±0.07	618±9.34	430±11.34	0.73±0.09	315±10.59	55±0.49	45±0.14	417±6.96
17	Jaisinghpura Bhuriya	8.98±0.01	1900±13.77	654±14.57	1.89±0.11	100±18.45	70±1.67	350±3.89	2100±13.45

18	Chimanpura	9.34±0.06	454±18.32	321±10.34	5.98±0.06	40± 15.98	10±1.39	55±1.58	516±7.65
19	Pratapnagar	9.5±0.04	410±9.77	453± 16.65	6.42±0.15	70±14.87	40±0.78	450±6.09	910±14.12
20	Mansrover	7.89±0.04	650± 12.34	1400±15.33	1.69±0.03	140±7.38	60±1.95	120±1.67	450±17.91
21	Maheshnagar	7.56±0.12	1210±16.76	2310±18.23	3.89±0.14	200±9.47	180±1.90	200±0.56	389±7.98
22	Manpura	8.78±0.08	1345±15.88	2365±13.09	2.43±0.16	272±10.69	118±0.68	210±2.55	980±11.45
23	Jhalana Doongri	8.27±0.01	879± 11.66	2215± 17.09	4.89±0.09	300±15.09	60±1.76	215±3.09	410±10.71

Table 1. Water Sample collected from Various Area of Sanganer Tehsil and analyzed

Table 2: Drinking Water Parameter is given by WHO, USPH, BIS

Parameter	USPH standard	WHOstandard	BIS standard
Color	Colorless	-	5
Odor	Odorless	-	Unobjectionable
Taste	Tasteless	-	Agreeable
pH	6.0-8.5	6.5-9.2	6.5-8.5
DO	4.0-6.0	-	3.0
TDS	500	500	500
TSS			100
CI	250	500	250
SO ₄	250	200	200
NO ₃	<10	45	45
so ₄ ⁻	-	-	400
F	1.5	0.5	1.0
PO ₄	0.1		-
Ca	100	100	75

Mg	30	150	30
COD	4.0	10	-
B.O.D.	-	-	30

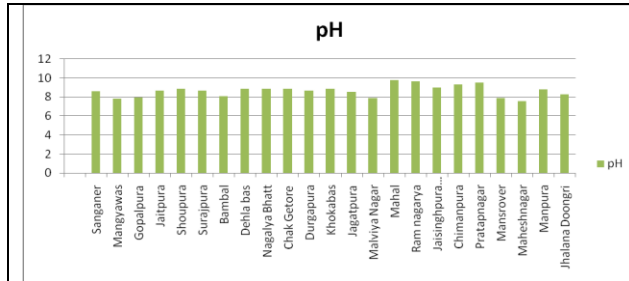


Fig.1 Showing pH of villages of Sanganer tehsil

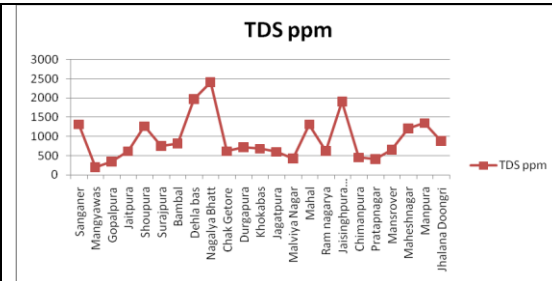


Fig.2 Showing T.D.S. of villages of Sanganer tehsil

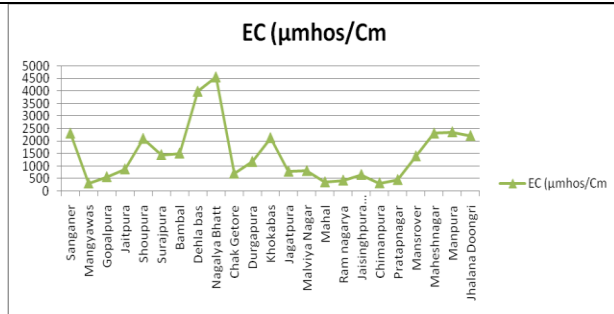


Fig.3 Showing E.C. of villages of Sanganer tehsil

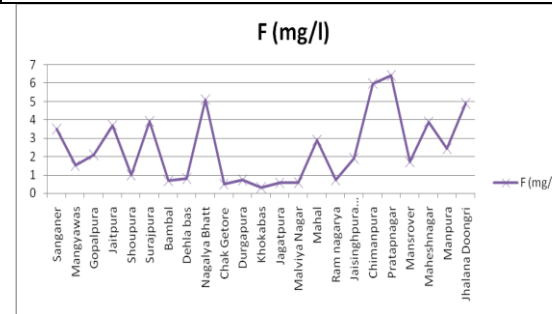


Fig.4. Showing fluoride of villages of Sanganer tehsil

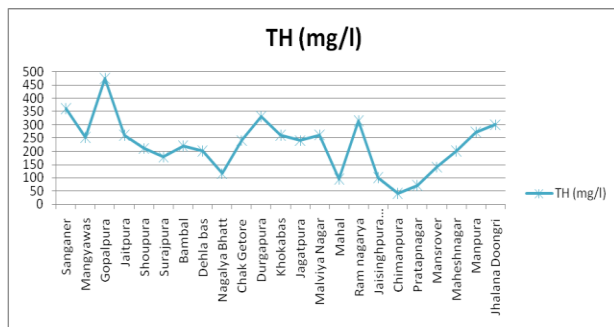


Fig.5 Showing Total Hardness of villages of Sanganer tehsil

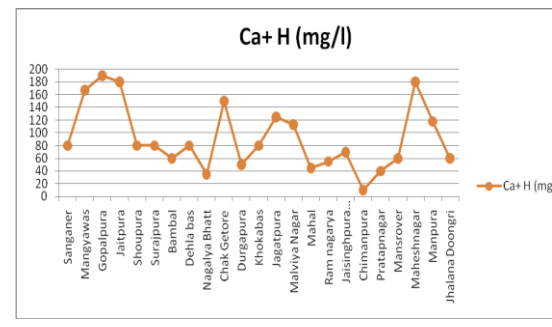


Fig.6 Showing Calcium Hardness of villages of Sanganer tehsil

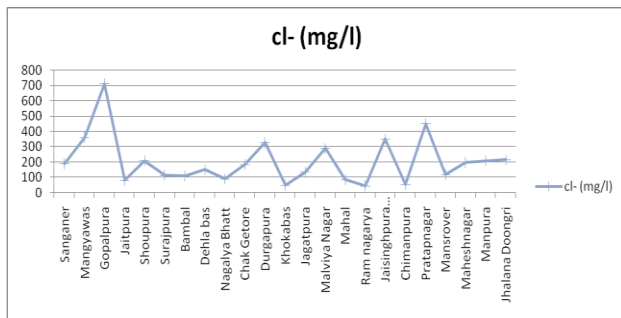


Fig.7. Showing Cl⁻ of villages of Sanganer tehsil

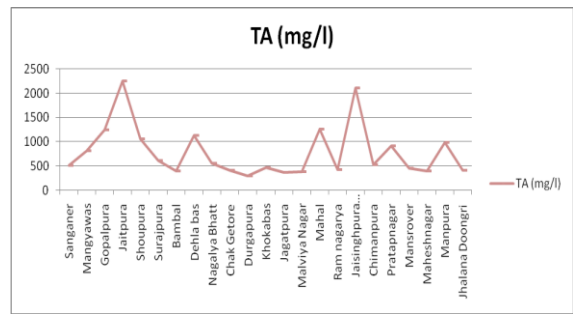


Fig.8. Showing Total Alkalinity of villages of Sanganer tehsil